- 7. Account for each of the following observations in terms of atomic theory and/or quantum theory.
  - (a) Atomic size decreases from Na to Cl in the periodic table.
  - (b) Boron commonly forms molecules of the type BX<sub>3</sub>. These molecules have a trigonal planar structure.
  - (b) Boron commonly forms molecules of the type  $BX_3$ . These molecules have a trigonal planar structure.

- ,
- (d) Each element displays a unique gas-phase emission spectrum.

(c) The first ionization energy of K is less than that of Na.